

1-(4-Methoxyphenyl)-2-methyl-1H-indole-3-carbonitrile

Qiao Yan and Xiuxiang Qi*

School of Pharmaceutical Science and Technology, Tianjin University, Tianjin 300072, People's Republic of China
 Correspondence e-mail: qxx@zzu.edu.cn

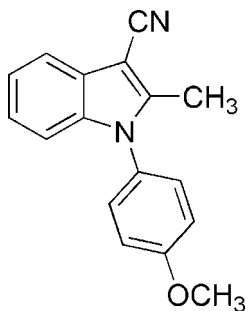
Received 22 July 2011; accepted 2 August 2011

Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.035; wR factor = 0.090; data-to-parameter ratio = 17.5.

In the title compound, $\text{C}_{17}\text{H}_{14}\text{N}_2\text{O}$, the dihedral angle between the indole ring system and the benzene ring is $58.41(4)^\circ$. The crystal packing features π - π stacking [shortest centroid-centroid separation = $3.8040(9)$ Å] and $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For the synthesis of the title compound, see: Du *et al.* (2006). For its precursor, see: Jin *et al.* (2009). For a related structure, see: Yang *et al.* (2011).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{14}\text{N}_2\text{O}$
 $M_r = 262.30$

Triclinic, $P\bar{1}$
 $a = 7.7381(10)$ Å

$b = 9.4598(14)$ Å
 $c = 9.7976(16)$ Å
 $\alpha = 95.983(2)^\circ$
 $\beta = 95.464(4)^\circ$
 $\gamma = 106.295(5)^\circ$
 $V = 678.79(17)$ Å³

$Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 113$ K
 $0.20 \times 0.18 \times 0.16$ mm

Data collection

Rigaku Saturn724 CCD diffractometer
 Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2009)
 $T_{\min} = 0.984$, $T_{\max} = 0.987$

8580 measured reflections
 3210 independent reflections
 2146 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.090$
 $S = 1.01$
 3210 reflections

183 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.19$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.24$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$\text{Cg}2$ and $\text{Cg}3$ are the centroids of the $\text{C}2-\text{C}7$ and $\text{C}8-\text{C}13$ rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}3-\text{H}3\cdots\text{Cg}3^{\text{i}}$	0.95	2.83	3.6542 (14)	146
$\text{C}10-\text{H}10\cdots\text{Cg}2^{\text{ii}}$	0.95	2.95	3.7133 (14)	138

Symmetry codes: (i) $x - 1, y, z$; (ii) $-x + 1, -y + 1, -z + 2$.

Data collection: *CrystalClear* (Rigaku, 2009); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2009); software used to prepare material for publication: *CrystalStructure*.

XQ is grateful for financial support for this project from the China Postdoctoral Science Foundation (200904507610).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6329).

References

- Du, Y., Liu, R., Linn, G. & Zhao, K. (2006). *Org. Lett.* **8**, 5919–5922.
 Jin, H., Li, P., Liu, B. & Cheng, X. (2009). *Acta Cryst.* **E65**, o236.
 Rigaku (2009). *CrystalClear* and *CrystalStructure*. Rigaku Corporation, Tokyo, Japan.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Yang, K., Li, P.-F., Liu, Y. & Fang, Z.-Z. (2011). *Acta Cryst.* **E67**, o1041.

supplementary materials

Acta Cryst. (2011). E67, o2312 [doi:10.1107/S1600536811031035]

1-(4-Methoxyphenyl)-2-methyl-1*H*-indole-3-carbonitrile

Q. Yan and X. Qi

Comment

In the molecular structure of the title compound, (I), (Fig. 1), the indole ring is almost planar with a dihedral angle of 2.66 (6)° between its pyrrole ring and fused benzene ring, which is greater than that [0.85 (6)°] of the 1-(2-chlorophenyl)-6-fluoro-2-methyl-1*H*-indole-3-carbonitrile reported by Yang *et al.* (2011). The indole ring constructs an angle of 58.41 (4)° with the methoxybenzene ring, which is much less than that [80.91 (5)°] reported by Yang *et al.* (2011).

In the crystal packing, π - π stacking interaction and C—H \cdots π interaction help establish the molecular packing. The shortest centroid-centroid separation is 3.8040 (9) Å, which occurs between the benzo part and pyrrole part of the molecules.

Experimental

The title compound was prepared according to the method of the literature (Du, *et al.*, 2006). Colourless prisms of (I) were grown from a mixture of ethyl acetate and petroleum ether.

Refinement

All H atoms were positioned geometrically (C—H = 0.95 and 0.98 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{CH})$ or $1.5U_{\text{eq}}(\text{CH}_3)$.

Figures

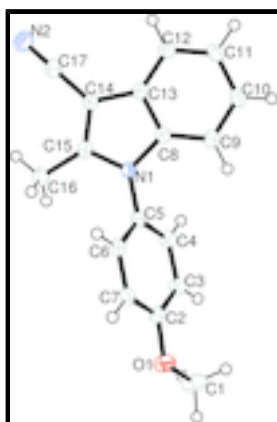


Fig. 1. The molecular structure of molecule one of (I) with 50% probability displacement ellipsoids.

1-(4-Methoxyphenyl)-2-methyl-1H-indole-3-carbonitrile

Crystal data

$C_{17}H_{14}N_2O$	$Z = 2$
$M_r = 262.30$	$F(000) = 276$
Triclinic, PT	$D_x = 1.283 \text{ Mg m}^{-3}$
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 7.7381 (10) \text{ \AA}$	Cell parameters from 2366 reflections
$b = 9.4598 (14) \text{ \AA}$	$\theta = 2.1\text{--}27.9^\circ$
$c = 9.7976 (16) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$\alpha = 95.983 (2)^\circ$	$T = 113 \text{ K}$
$\beta = 95.464 (4)^\circ$	Prism, colorless
$\gamma = 106.295 (5)^\circ$	$0.20 \times 0.18 \times 0.16 \text{ mm}$
$V = 678.79 (17) \text{ \AA}^3$	

Data collection

Rigaku Saturn724 CCD diffractometer	3210 independent reflections
Radiation source: rotating anode multilayer	2146 reflections with $I > 2\sigma(I)$
Detector resolution: $14.22 \text{ pixels mm}^{-1}$	$R_{\text{int}} = 0.035$
ω and ϕ scans	$\theta_{\text{max}} = 27.9^\circ$, $\theta_{\text{min}} = 2.1^\circ$
Absorption correction: multi-scan (<i>CrystalClear</i> ; Rigaku, 2009)	$h = -10 \rightarrow 10$
$T_{\text{min}} = 0.984$, $T_{\text{max}} = 0.987$	$k = -12 \rightarrow 12$
8580 measured reflections	$l = -12 \rightarrow 12$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.035$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.090$	H-atom parameters constrained
$S = 1.01$	$w = 1/[\sigma^2(F_o^2) + (0.038P)^2]$
3210 reflections	where $P = (F_o^2 + 2F_c^2)/3$
183 parameters	$(\Delta/\sigma)_{\text{max}} = 0.001$
0 restraints	$\Delta\rho_{\text{max}} = 0.19 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	1.23428 (10)	0.89022 (9)	0.06414 (8)	0.0300 (2)
N1	0.63352 (12)	0.69683 (10)	0.34222 (9)	0.0207 (2)
N2	0.27094 (14)	0.77814 (11)	0.68887 (10)	0.0350 (3)
C1	1.34704 (16)	0.79705 (15)	0.03434 (14)	0.0386 (3)
H1A	1.3990	0.7732	0.1211	0.058*
H1B	1.4452	0.8492	-0.0148	0.058*
H1C	1.2741	0.7050	-0.0237	0.058*
C2	1.08888 (14)	0.83382 (12)	0.13188 (11)	0.0223 (3)
C3	1.04548 (15)	0.69409 (12)	0.17549 (11)	0.0246 (3)
H3	1.1173	0.6293	0.1573	0.030*
C4	0.89572 (14)	0.65052 (12)	0.24598 (11)	0.0241 (3)
H4	0.8661	0.5558	0.2771	0.029*
C5	0.78939 (14)	0.74363 (12)	0.27134 (11)	0.0207 (2)
C6	0.83139 (14)	0.88247 (12)	0.22493 (11)	0.0221 (2)
H6	0.7570	0.9457	0.2400	0.026*
C7	0.98146 (14)	0.92716 (12)	0.15712 (11)	0.0226 (2)
H7	1.0119	1.0225	0.1274	0.027*
C8	0.48670 (14)	0.57024 (11)	0.29653 (11)	0.0199 (2)
C9	0.45885 (14)	0.46469 (12)	0.17956 (11)	0.0239 (3)
H9	0.5482	0.4695	0.1186	0.029*
C10	0.29541 (15)	0.35277 (13)	0.15628 (12)	0.0279 (3)
H10	0.2715	0.2794	0.0774	0.033*
C11	0.16419 (15)	0.34554 (12)	0.24707 (12)	0.0273 (3)
H11	0.0531	0.2676	0.2283	0.033*
C12	0.19364 (15)	0.44962 (12)	0.36305 (12)	0.0235 (3)
H12	0.1051	0.4429	0.4248	0.028*
C13	0.35650 (14)	0.56517 (12)	0.38782 (11)	0.0205 (2)
C14	0.42996 (15)	0.69530 (12)	0.48926 (11)	0.0216 (2)
C15	0.59797 (14)	0.77272 (12)	0.45929 (11)	0.0213 (2)
C16	0.72912 (15)	0.91064 (12)	0.53687 (11)	0.0293 (3)
H16A	0.7311	0.9943	0.4852	0.044*
H16B	0.8506	0.8976	0.5486	0.044*
H16C	0.6920	0.9310	0.6280	0.044*

supplementary materials

C17 0.34392 (15) 0.74071 (12) 0.60128 (12) 0.0246 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0284 (4)	0.0321 (5)	0.0346 (5)	0.0114 (4)	0.0121 (4)	0.0123 (4)
N1	0.0221 (5)	0.0197 (5)	0.0199 (5)	0.0056 (4)	0.0027 (4)	0.0018 (4)
N2	0.0394 (6)	0.0346 (6)	0.0341 (6)	0.0142 (5)	0.0124 (5)	0.0027 (5)
C1	0.0320 (7)	0.0475 (8)	0.0473 (8)	0.0209 (6)	0.0166 (6)	0.0203 (7)
C2	0.0222 (6)	0.0251 (6)	0.0184 (6)	0.0054 (5)	0.0012 (5)	0.0034 (5)
C3	0.0247 (6)	0.0267 (6)	0.0265 (6)	0.0128 (5)	0.0033 (5)	0.0070 (5)
C4	0.0266 (6)	0.0217 (6)	0.0255 (6)	0.0086 (5)	0.0018 (5)	0.0073 (5)
C5	0.0208 (5)	0.0225 (6)	0.0179 (5)	0.0057 (5)	0.0006 (4)	0.0024 (4)
C6	0.0259 (6)	0.0199 (6)	0.0203 (6)	0.0085 (5)	0.0002 (5)	0.0000 (5)
C7	0.0282 (6)	0.0178 (6)	0.0200 (6)	0.0048 (5)	0.0010 (5)	0.0019 (4)
C8	0.0208 (5)	0.0183 (5)	0.0212 (6)	0.0072 (4)	0.0000 (4)	0.0041 (4)
C9	0.0268 (6)	0.0237 (6)	0.0226 (6)	0.0098 (5)	0.0032 (5)	0.0024 (5)
C10	0.0307 (6)	0.0223 (6)	0.0287 (7)	0.0077 (5)	0.0005 (5)	-0.0022 (5)
C11	0.0238 (6)	0.0201 (6)	0.0361 (7)	0.0055 (5)	0.0001 (5)	0.0016 (5)
C12	0.0234 (6)	0.0219 (6)	0.0284 (6)	0.0098 (5)	0.0050 (5)	0.0067 (5)
C13	0.0237 (6)	0.0193 (5)	0.0209 (6)	0.0104 (5)	0.0011 (5)	0.0039 (4)
C14	0.0261 (6)	0.0216 (6)	0.0195 (6)	0.0107 (5)	0.0027 (4)	0.0038 (5)
C15	0.0262 (6)	0.0203 (6)	0.0184 (6)	0.0091 (5)	0.0014 (4)	0.0023 (4)
C16	0.0339 (6)	0.0258 (6)	0.0243 (6)	0.0046 (5)	0.0023 (5)	-0.0006 (5)
C17	0.0279 (6)	0.0213 (6)	0.0258 (6)	0.0090 (5)	0.0028 (5)	0.0040 (5)

Geometric parameters (\AA , $^\circ$)

O1—C2	1.3675 (12)	C7—H7	0.9500
O1—C1	1.4317 (13)	C8—C9	1.3965 (14)
N1—C15	1.3805 (13)	C8—C13	1.4038 (14)
N1—C8	1.3976 (13)	C9—C10	1.3842 (15)
N1—C5	1.4342 (13)	C9—H9	0.9500
N2—C17	1.1497 (13)	C10—C11	1.4045 (15)
C1—H1A	0.9800	C10—H10	0.9500
C1—H1B	0.9800	C11—C12	1.3799 (15)
C1—H1C	0.9800	C11—H11	0.9500
C2—C7	1.3919 (15)	C12—C13	1.3984 (15)
C2—C3	1.3923 (15)	C12—H12	0.9500
C3—C4	1.3905 (15)	C13—C14	1.4391 (15)
C3—H3	0.9500	C14—C15	1.3770 (15)
C4—C5	1.3837 (14)	C14—C17	1.4261 (15)
C4—H4	0.9500	C15—C16	1.4877 (15)
C5—C6	1.3966 (14)	C16—H16A	0.9800
C6—C7	1.3779 (14)	C16—H16B	0.9800
C6—H6	0.9500	C16—H16C	0.9800
C2—O1—C1	117.18 (9)	N1—C8—C13	108.14 (9)
C15—N1—C8	109.22 (9)	C10—C9—C8	117.08 (10)

C15—N1—C5	126.23 (9)	C10—C9—H9	121.5
C8—N1—C5	124.42 (9)	C8—C9—H9	121.5
O1—C1—H1A	109.5	C9—C10—C11	121.28 (10)
O1—C1—H1B	109.5	C9—C10—H10	119.4
H1A—C1—H1B	109.5	C11—C10—H10	119.4
O1—C1—H1C	109.5	C12—C11—C10	121.21 (11)
H1A—C1—H1C	109.5	C12—C11—H11	119.4
H1B—C1—H1C	109.5	C10—C11—H11	119.4
O1—C2—C7	115.32 (10)	C11—C12—C13	118.65 (11)
O1—C2—C3	124.61 (10)	C11—C12—H12	120.7
C7—C2—C3	120.07 (10)	C13—C12—H12	120.7
C4—C3—C2	119.16 (10)	C12—C13—C8	119.40 (10)
C4—C3—H3	120.4	C12—C13—C14	134.75 (10)
C2—C3—H3	120.4	C8—C13—C14	105.80 (10)
C5—C4—C3	120.69 (10)	C15—C14—C17	124.72 (10)
C5—C4—H4	119.7	C15—C14—C13	108.77 (9)
C3—C4—H4	119.7	C17—C14—C13	126.50 (10)
C4—C5—C6	119.91 (10)	C14—C15—N1	108.06 (10)
C4—C5—N1	120.31 (10)	C14—C15—C16	128.79 (10)
C6—C5—N1	119.76 (10)	N1—C15—C16	123.11 (10)
C7—C6—C5	119.59 (10)	C15—C16—H16A	109.5
C7—C6—H6	120.2	C15—C16—H16B	109.5
C5—C6—H6	120.2	H16A—C16—H16B	109.5
C6—C7—C2	120.56 (10)	C15—C16—H16C	109.5
C6—C7—H7	119.7	H16A—C16—H16C	109.5
C2—C7—H7	119.7	H16B—C16—H16C	109.5
C9—C8—N1	129.45 (10)	N2—C17—C14	178.02 (12)
C9—C8—C13	122.36 (10)		
C1—O1—C2—C7	-179.16 (9)	C9—C10—C11—C12	-0.16 (17)
C1—O1—C2—C3	1.16 (16)	C10—C11—C12—C13	1.13 (17)
O1—C2—C3—C4	178.73 (9)	C11—C12—C13—C8	-1.51 (16)
C7—C2—C3—C4	-0.94 (16)	C11—C12—C13—C14	175.65 (11)
C2—C3—C4—C5	0.76 (16)	C9—C8—C13—C12	0.98 (16)
C3—C4—C5—C6	0.52 (16)	N1—C8—C13—C12	178.74 (9)
C3—C4—C5—N1	179.03 (9)	C9—C8—C13—C14	-176.92 (9)
C15—N1—C5—C4	125.90 (12)	N1—C8—C13—C14	0.84 (11)
C8—N1—C5—C4	-58.75 (14)	C12—C13—C14—C15	-178.10 (11)
C15—N1—C5—C6	-55.59 (14)	C8—C13—C14—C15	-0.67 (12)
C8—N1—C5—C6	119.76 (11)	C12—C13—C14—C17	0.6 (2)
C4—C5—C6—C7	-1.62 (15)	C8—C13—C14—C17	178.00 (10)
N1—C5—C6—C7	179.86 (9)	C17—C14—C15—N1	-178.46 (10)
C5—C6—C7—C2	1.45 (16)	C13—C14—C15—N1	0.24 (12)
O1—C2—C7—C6	-179.86 (9)	C17—C14—C15—C16	3.84 (18)
C3—C2—C7—C6	-0.16 (16)	C13—C14—C15—C16	-177.46 (10)
C15—N1—C8—C9	176.83 (10)	C8—N1—C15—C14	0.29 (12)
C5—N1—C8—C9	0.80 (17)	C5—N1—C15—C14	176.23 (9)
C15—N1—C8—C13	-0.72 (12)	C8—N1—C15—C16	178.15 (9)
C5—N1—C8—C13	-176.75 (9)	C5—N1—C15—C16	-5.91 (16)
N1—C8—C9—C10	-177.26 (10)	C15—C14—C17—N2	110 (4)

supplementary materials

C13—C8—C9—C10	-0.01 (16)	C13—C14—C17—N2	-68 (4)
C8—C9—C10—C11	-0.40 (16)		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg2 and Cg3 are the centroids of the C2–C7 and C8–C13 rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 \cdots Cg3 ⁱ	0.95	2.83	3.6542 (14)	146
C10—H10 \cdots Cg2 ⁱⁱ	0.95	2.95	3.7133 (14)	138

Symmetry codes: (i) $x-1, y, z$; (ii) $-x+1, -y+1, -z+2$.

Fig. 1

